

QUANTITATIVE CHEMICAL ANALYSIS



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3 LE CHATELIER S PRINCIPLE $\text{NaCl(s)} + \text{H}_2\text{SO}_4\text{(l)} \rightarrow \text{Na}_2\text{SO}_4\text{(s)} + \text{HCl(g)}$ reaction is carried out in an open container Because HCl is a gas that can escape from the system,

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Biographical Statement for Nomination of Daniel C. Harris for J. Calvin Giddings Award for Excellence in Analytical Chemical Education I was born in 1948 in Brooklyn, New York.

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